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CHAPTER 5 REVIEW The Periodic Law SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number. (d) based on when they were discovered. 2. d Mendeleev noticed that certain similarities in the ...

5 The Periodic Law - Jefferson Township Public Schools

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a mutual electrical attraction between the nuclei and the valence electrons of different atoms that binds the atoms together

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CHAPTER 10 REVIEW States of Matter SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Match description on the right to the correct crystal type on the left. b ionic crystal (a) has mobile electrons in the crystal c covalent molecular crystal (b) is hard, brittle, and nonconducting a metallic crystal (c) typically has the lowest melting point of the four

10 States of Matter - Ms. Agostine's Chemistry Page

The rest of the pages are the exact same as the non state specific study guide. So if your student uses the HMH Modern Chemistry Edition (2017) this book will help further their learning. It does not come with the answers. The answers are found within the teacher resource material for the HMH Modern Chemistry (2017) curriculum.

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